#### **Quantum Monte Carlo for Noncovalent Interactions**

Advances in quantum Monte Carlo techniques for non-relativistic many-body systems (June 28, 2013, Seattle)

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**Collaboration**:

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# **Outline & Scope**

- Numerical evidence
  - FN-DMC in NCI what can be achieved?

// No method development, skip details on VMC & DMC & technical details //

FUNDAMENTAL QUESTIONS OF THIS MEETING ADRESSED: What is the current state of the art of QMC as compared to other many-body techniques? Is it possible to imagine QMC as a reliable standard to be used also by non- experts in the near future? What is the cost/benefit ratio compared, for instance, to DFT calculations?

http://www.int.washington.edu/PROGRAMS/13-2a/Questions.html

# Noncovalent Molecular Interactions

- Chemical <u>bonding without sharing of electrons</u>
  - Hydrogen bonds
  - Van der Waals
    - Dipole-dipole, London dispersion
- Importance
  - Structure of biomacromolecules
  - Properties of liquids
  - Molecular recognition
  - •

#### Typical strength: 0.5-30 kcal/mol

# Levels of accuracy

- Depends on the problem
- Chemical accuracy
  - 1 kcal/mol ~ 0.04 eV
- Scale of NCI starts at 0.5 kcal/mol
  - Need less than 1 kcal/mol
- Subchemical accuracy: 0.1 kcal/mol
  - Target benchmark level for NCI
  - One of the most challenging tasks in computational chemisty

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#### **OUR GOAL!**

# **Noncovalent Interactions**

#### • Experiment

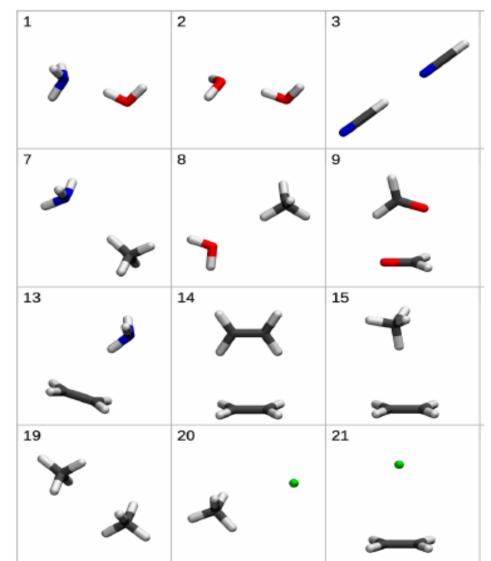
- Strength on interactions
  - Dissociation & adsorption enthalpies
  - No direct info on nature of interactions
- Theory
  - Enthalpy
    - hard for anharmonicity
  - Interaction energy available from SSE in BO approx.
  - Other quantities of interest fundamental understanding

# Theory

- Problem: solve SNR-SE in BO approximation
- Golden standard of QCH: CCSD(T)
- Accurate in large basis and/or in CBS limit  $\rightarrow 1$  kcal/mol
- Problem: rapid scaling with system complexity ~O(N^7)
- Is the CCSD(T)/CBS reliable? How much?
- In general this is not settled
  - Single reference method! OK for NCI

#### Recent assessment of CCSD(T) on small complexes

- Rezac, Hobza JCTC 2013
  - Tests of various approximations on noncovalent interactions, test set A24:
    - Relativity
    - Excitation order by CCSDT(Q)
    - Ignoring core-valence correlation



	system		CCSD(T)/CBS	<b>ACCSD</b>	Γ(Q) (%)
1	water…ammonia	Cs	-6.493	0.001	(0.01)
2	water dimer	$C_s$	-5.006	0.012	(0.23)
3	HCN dimer	$C_s$	-4.745	0.007	(0.15)
4	HF dimer	$C_s$	-4.581	0.017	(0.38)
5	ammonia dimer	$C_{2h}$	-3.137	-0.004	(0.13)
6	HF…methane	$C_{3v}$	-1.654	-0.006	(0.37)
7	ammonia…methane	$C_{3v}$	-0.765	-0.006	(0.80)

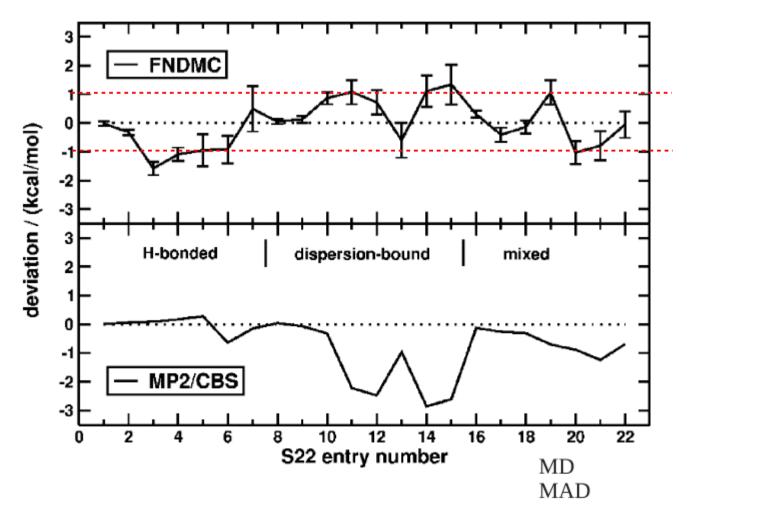
- Subchemical accuracy is achieved within the whole set >20 mol's, total avg. error on IE is 1.5 % only
- Error compensation
- CCSD(T) is "converged" for IEs in small noncovalent complexes our reference

# What about Quantum Monte Carlo?

- First step, assess FN-DMC w.r.t. CCSD(T) in small complexes
  - Then test on larges systems
  - Learn what's possible: goal
- Chemistry: screening of large sets
  - Feasible and black-box approach required: goal
- How well are we able to reach the benchmark CCSD(T)/CBS data on small molecules?

# Previous QMC attempt on a set S22

 Chemists are not satisfied with ~1 kcal/mol average error reported by Korth et al. JPCA 2008

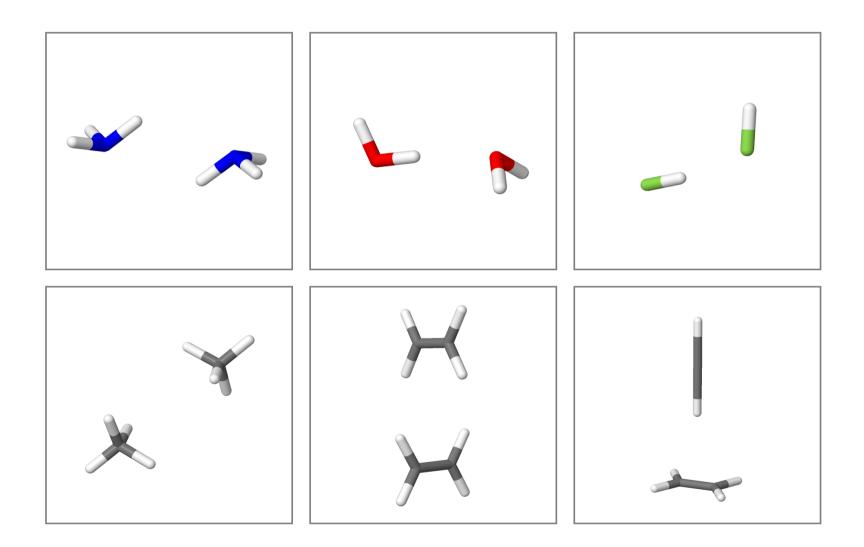


-0.03

0.68

#### Our work...

#### Test set



Complex	Reference	$\mathrm{FN}\text{-}\mathrm{DMC}^a$	$\Delta^a$	$\mathrm{FN} ext{-}\mathrm{DMC}^b$	$\Delta^b$
Ammonia dimer	-3.15	$-3.19 {\pm} 0.09$	0.04	$-3.22 \pm 0.10$	0.07
Water dimer	-5.07	$-5.34 {\pm} 0.09$	0.27	$-5.15 \pm 0.10$	0.08
HF dimer	-4.58	-	-	$-4.68 {\pm} 0.10$	0.10
Methane dimer	-0.53	$-0.48 {\pm} 0.08$	-0.05	$-0.44 \pm 0.10$	-0.09
Ethene dimer	-1.48	$-1.38 {\pm} 0.13$	-0.10	$-1.47 {\pm} 0.09$	-0.01
Ethene/ethyne	-1.50	$-1.22 \pm 0.12$	-0.28	$-1.56 \pm 0.10$	0.06

Complex	Reference	$\mathbf{I}$ N-DMC <sup>a</sup>	$\Delta^a$	$\mathrm{FN}\text{-}\mathrm{DMC}^{b}$	$\Delta^b$
Ammonia dimer	-3.15	- <b>B</b> .19±0.09	0.04	$-3.22 \pm 0.10$	0.07
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		-			

CCSD(T)/CBS ATZV - AQZV Takatani et al. JCP 2010

				-
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Korth et al. JPCA 2009

ME: -0.008 MUE: 0.116

#### **OUR DATA**

				I.	
Complex	Reference	$FN-DMC^{a}$	$\Delta^a$	$FN-DMC^b$	$\Delta^b$
Ammonia dimer	-3.15	$-3.19 {\pm} 0.09$	0.04	$-3.22 \pm 0.10$	0.07
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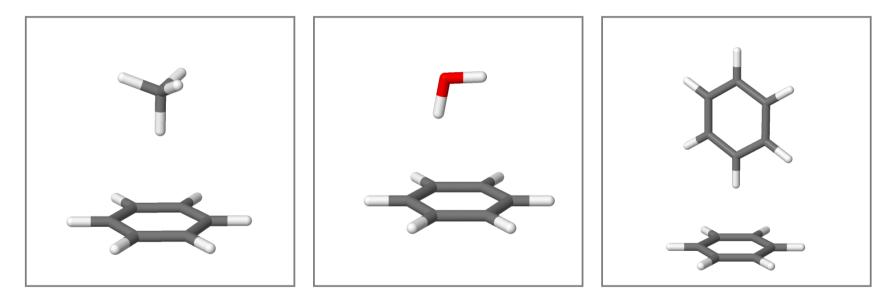
Korth et al. JPCA 2009

ME: -0.008 ME: 0.035 MUE: 0.116 MUE: 0.068 I.e. FN-DMC agrees to within subchemical accuracy w.r.t. benchmark data believed to be (esentially) exact.

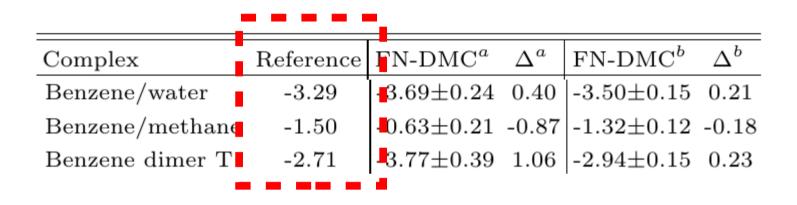
This makes FN-DMC competitor of CCSD(T) and in large complexes, it will benefit from the scaling.

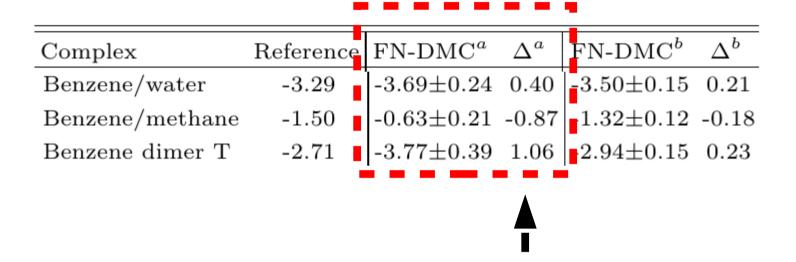
#### Larger complexes

CCSD(T) not yet assessed! Just best energy estimates... believed to be OK ~+-0.3 kcal/mol



Complex	Reference	$FN-DMC^{a}$	$\Delta^a$	$\mathrm{FN} ext{-}\mathrm{DMC}^b$	$\Delta^b$
Benzene/water	-3.29	$-3.69 {\pm} 0.24$	0.40	$-3.50 \pm 0.15$	0.21
Benzene/methane				$-1.32 \pm 0.12$	
Benzene dimer T	-2.71	$-3.77 {\pm} 0.39$	1.06	$-2.94{\pm}0.15$	0.23





Korth et al. JPCA 2009

#### MUE: 0.76

#### OUR DATA

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Benzene/methane	-1.50	$-0.63 \pm 0.21$	-0.87	$-1.32 \pm 0.12$	-0.18
Benzene dimer T	-2.71	$-3.77 {\pm} 0.39$	1.06	$-2.94{\pm}0.15$	0.23

Korth et al. JPCA 2009

#### MUE: 0.76 MUE: 0.213

# **Optimal Protocol**

- Geometries from S22 (except HF dimer)
- BFD ECP's
- Augmented bases TZV aug part is a must!
- B3LYP orbitals (no orbital opt)
- VMC opt of J only, 3 body Schmidt-Moskowitz, Poly pade
  - Linear combination of Energy & Variance
    - esentially energy minimization / 95% of energy
- DMC: T-moves, conservative dt=0.005 a.u.,
  - 0.1 kcal/mol or smaller error bar

#### Just accepted in JCTC http://pubs.acs.org/doi/abs/10.1021/ct4006739

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  - Finite variance, one determinant,...
- Energy differences are converged

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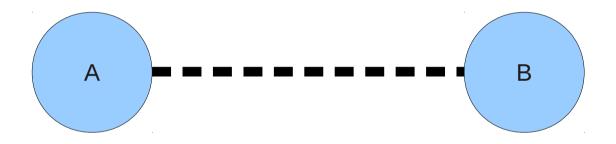
- The total energies are not converged,
  - Finite variance
- Energy differences are converged
- Why?
  - Efficient FN error cancellation
  - Closed shells no multireference nature of the wave functions arises upon dissociation of the molecular complexes constituents – equal footing description
  - Other...

#### **FN error cancellation**

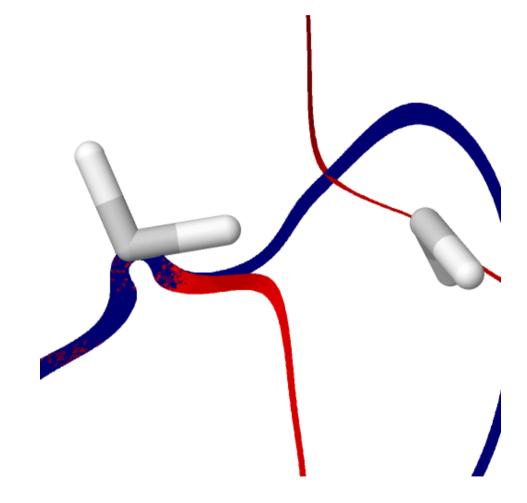
#### Anderson, Korth et al, ...

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			_	!		

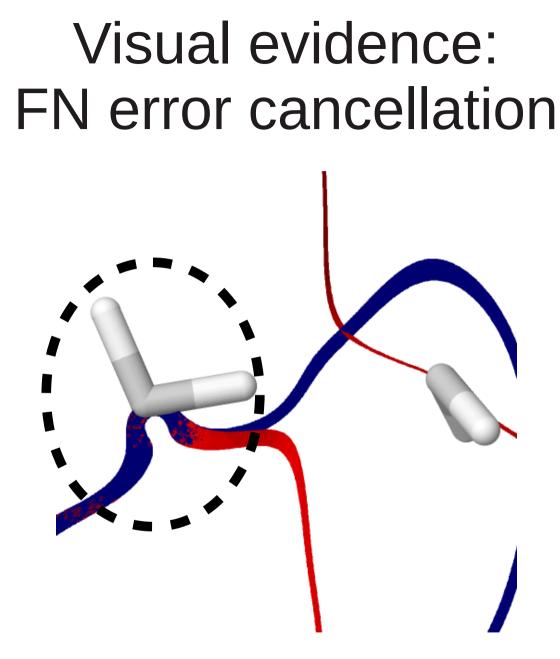
#### Weak interation => FN error constant, cancels out



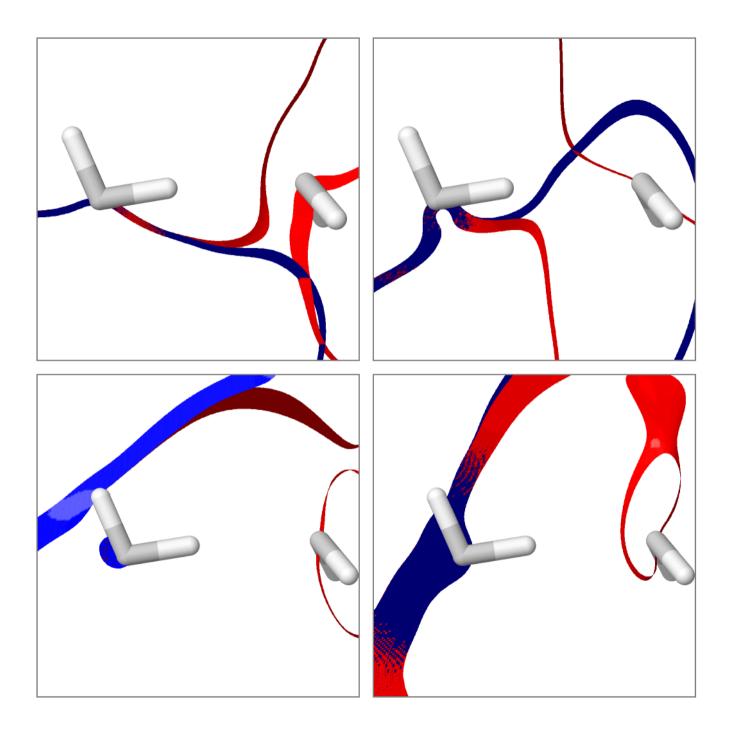
#### Visual evidence, water dimer Slice cuts through multidimensional nodal surface

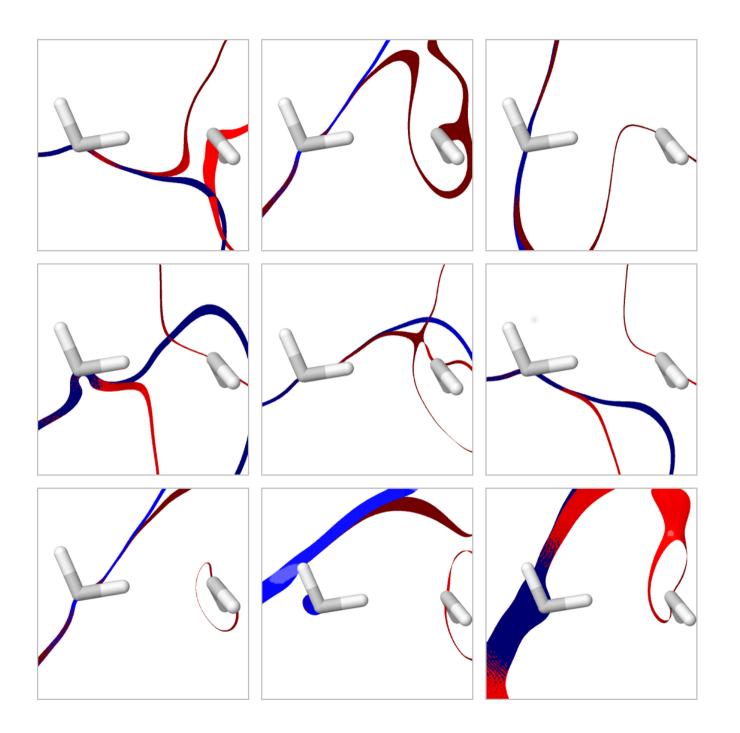


Monomer, dimer



Monomer, dimer





#### Too fresh... quantitative analysis under way...

#### FUNDAMENTAL QUESTIONS AGAIN

What is the current state of the art of QMC as compared to other many-body techniques? Is it possible to imagine QMC as a reliable standard to be used also by nonexperts in the near future? What is the cost/benefit ratio compared, for instance, to DFT calculations?

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# My answer in domain of NCI

- QMC now allows routine use & attains predictive power with benchmark accuracy as CCSD(T)
  - At least for comparable closed shell complexes with comparably complex bonding pattern/s
- I believe that NCI may be easily studied by non-QMC-expert using this approach as the provided protocol is esentially a black-box recipe
- More work required/under way, to support...
  - Testing on more complexes, (S22, cd, ...)
  - Predictive calculations
  - Physics nodes, nonlinearities?

#### Thank you!

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