

Surface Modification of CaF_2



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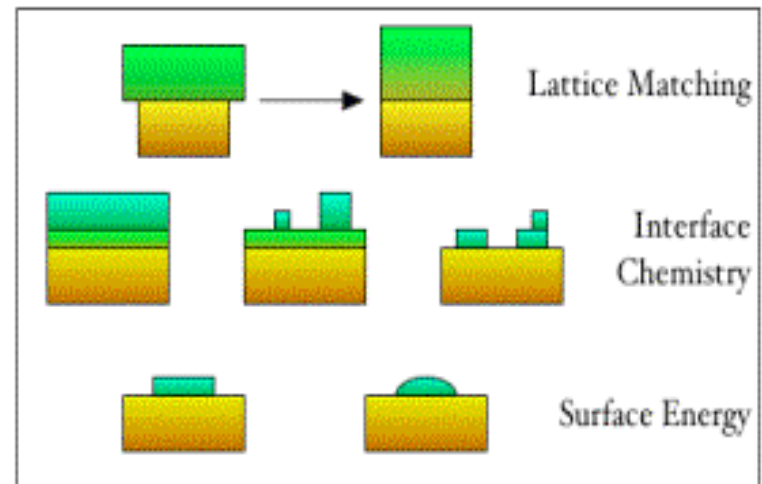
Outline



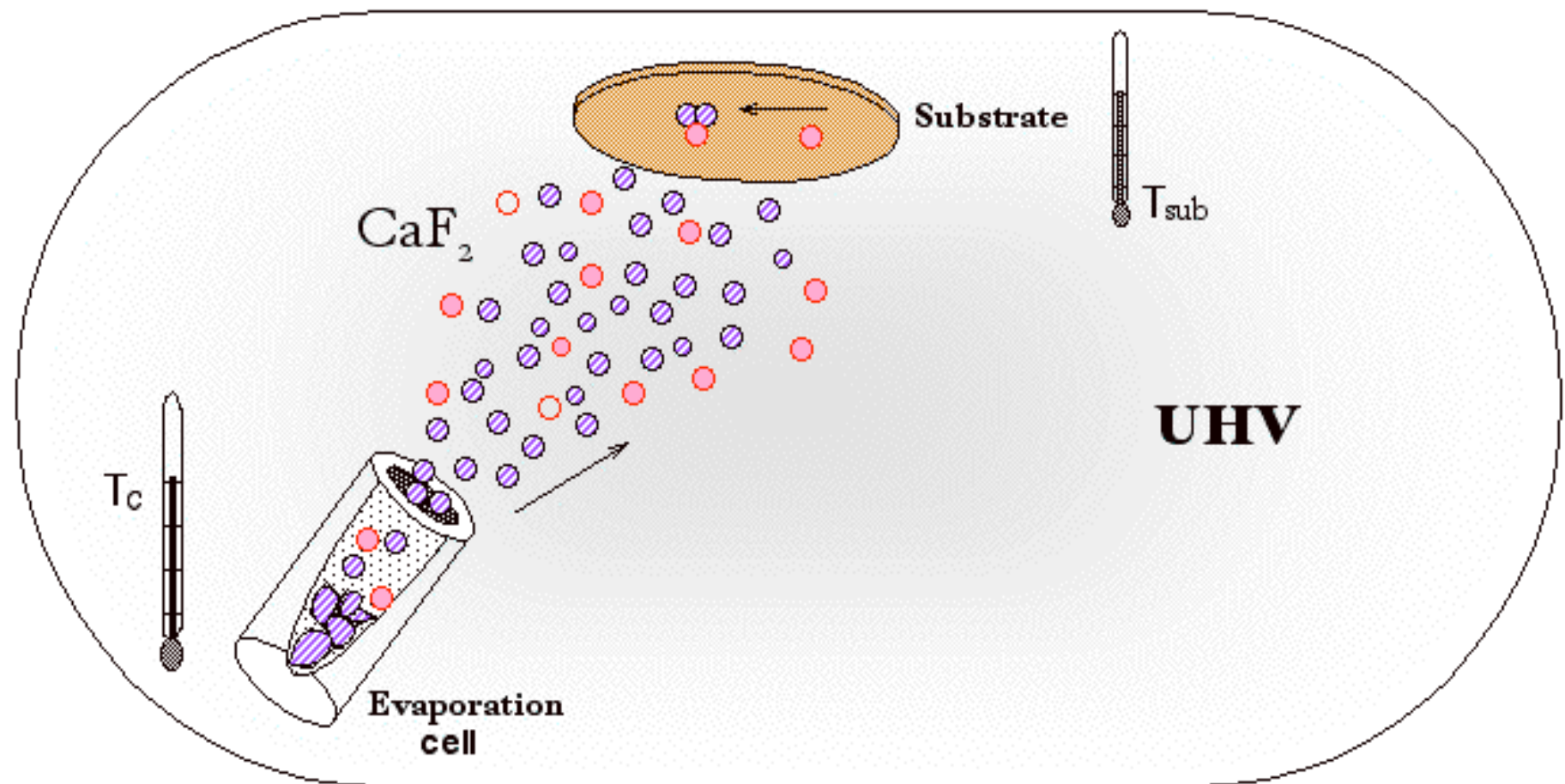
- Introduction
- Background
 - Molecular Beam Epitaxy
 - Apparatus
 - Analysis
- CaF_2
 - Electron Irradiation
 - As Termination

Introduction

- What is Epitaxy/ Epitaxial Growth?
 - The growth of one Crystal on another
- GaSe, AlSe, GaAs, TiO_2 , **CaF₂** on Si
- Various factors:
 - Lattice Match
 - Interface Chemistry
 - Surface Energy

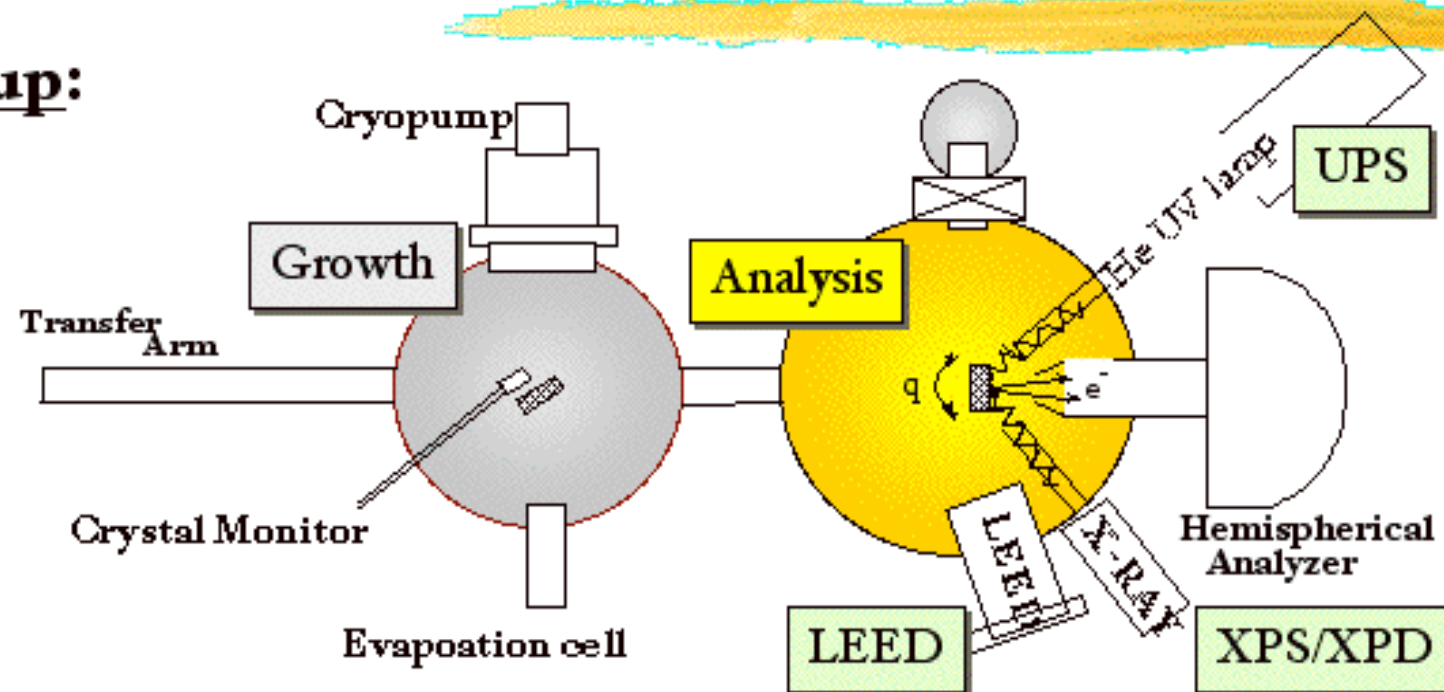


Molecular Beam Epitaxy



The Chamber

Setup:

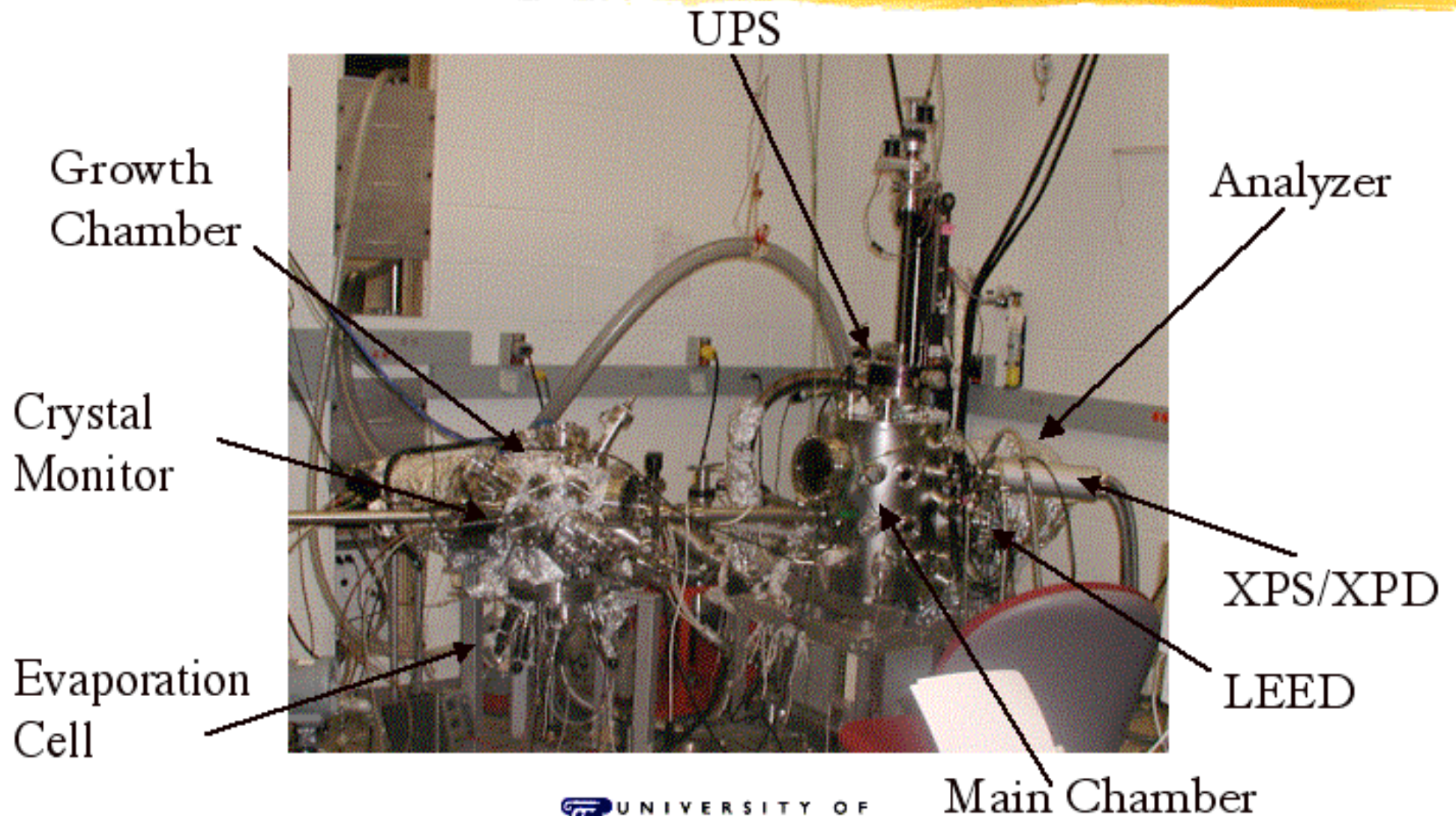


- Growth Control —
 - Substrate Temp, T_{sub}
 - Cell flux, Cell Temp.
 - Growth time



- Analysis —
 - LEED
 - XPS/XPD
 - UPS

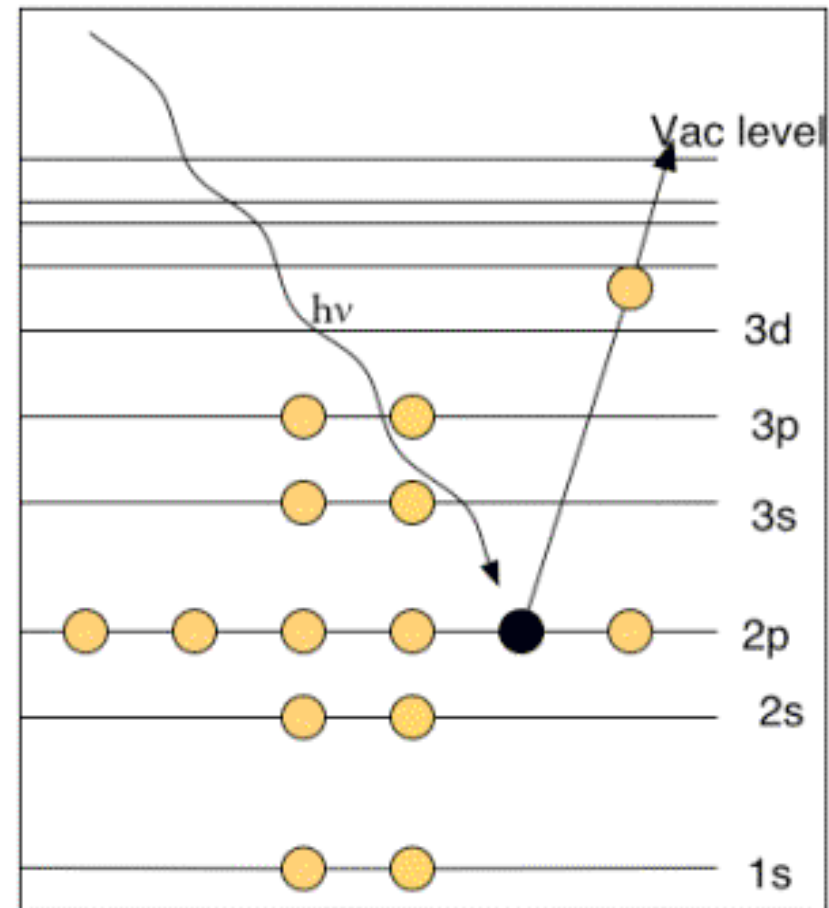
The Chamber



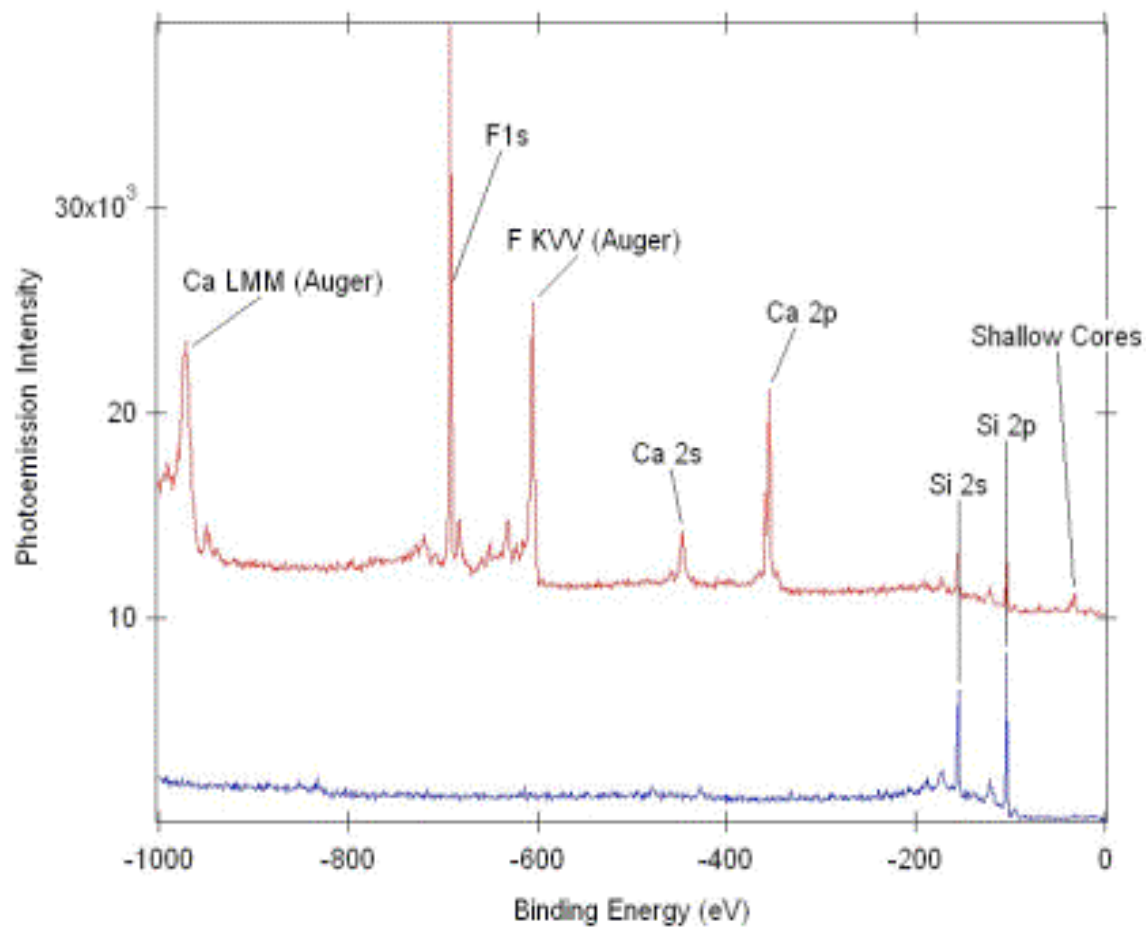
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XPS

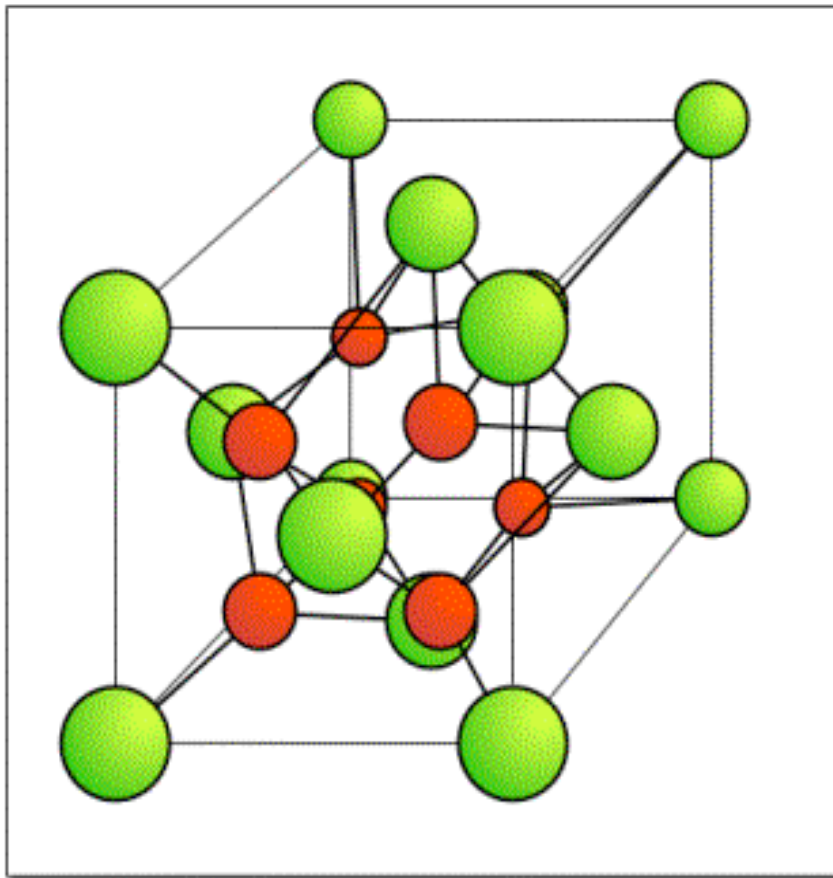
- Incident photons transfer energy to e^- in the crystal
- Photo-electron energies are scanned
- $KE(e^-) = h\nu - BE - \phi_{inner}$



XPS



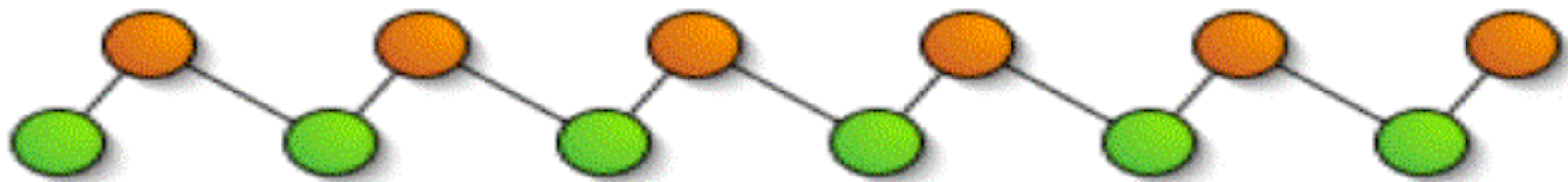
Why CaF_2 ?



- Ionic Crystal
- Insulator with a great lattice match to Si (0.6% mismatch)
- Quantum wells, Confinement

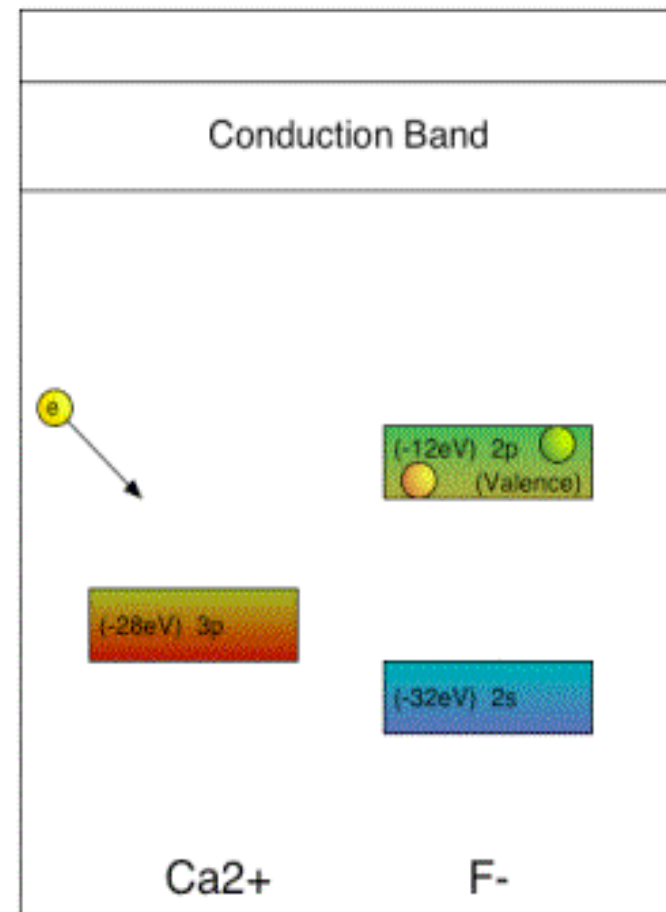
Surface of CaF_2

- Stable in UHV
- No surface reconstruction
- F^- termination is problematic
- Murphy's Law of Epitaxy
 - "If A grows well on B, then B doesn't grow well on A"
 - $\sigma(\text{Si}) > 300\% \sigma(\text{CaF}_2)$



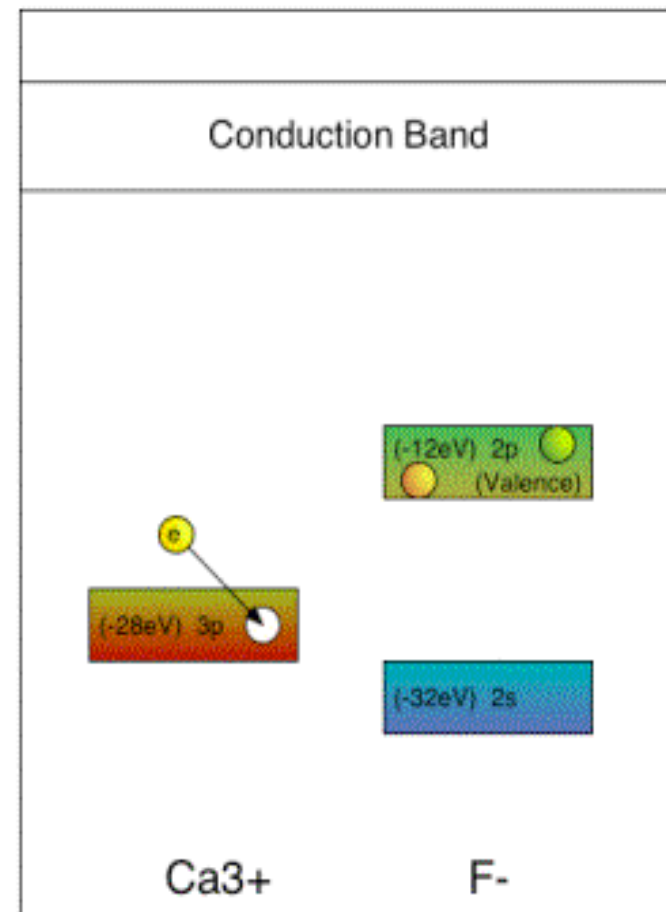
Knotek-Feibelman

- e^- of at least $\sim 40\text{eV}$ are “shot” at the sample



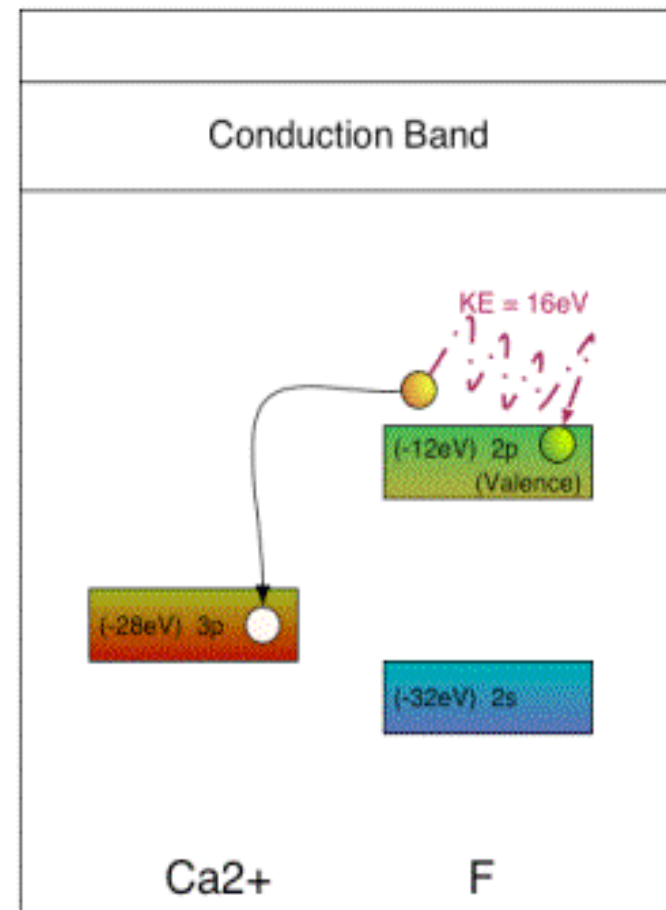
Knotek-Feibelman

- e^- of at least $\sim 40\text{eV}$ are “shot” at the sample
- A hole is created in the Ca 3p band



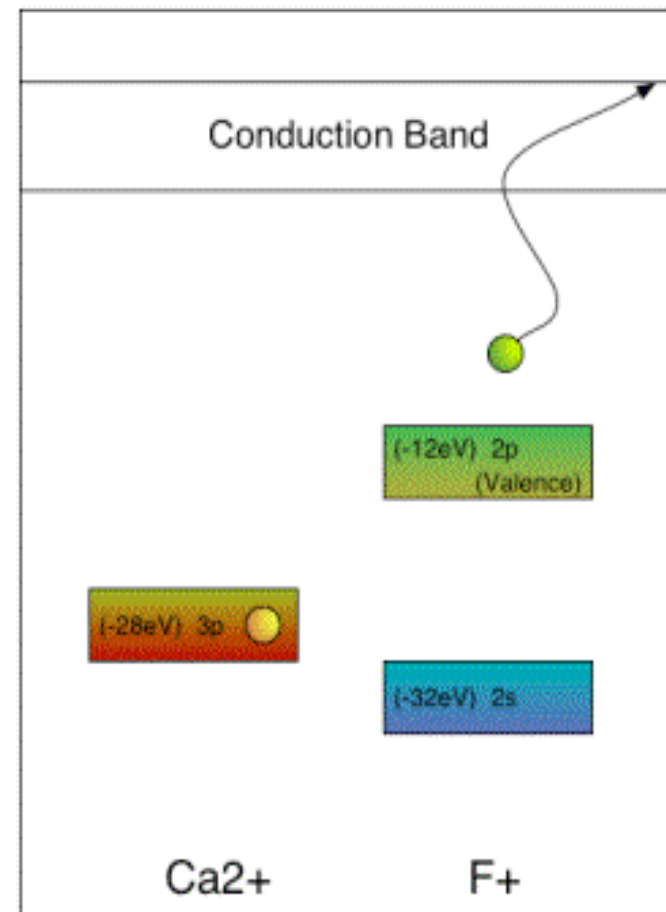
Knotek-Feibelman

- e^- of at least $\sim 40\text{eV}$ are “shot” at the sample
- A hole is created in the Ca 3p band
- An e^- from the F 2p decays into the hole

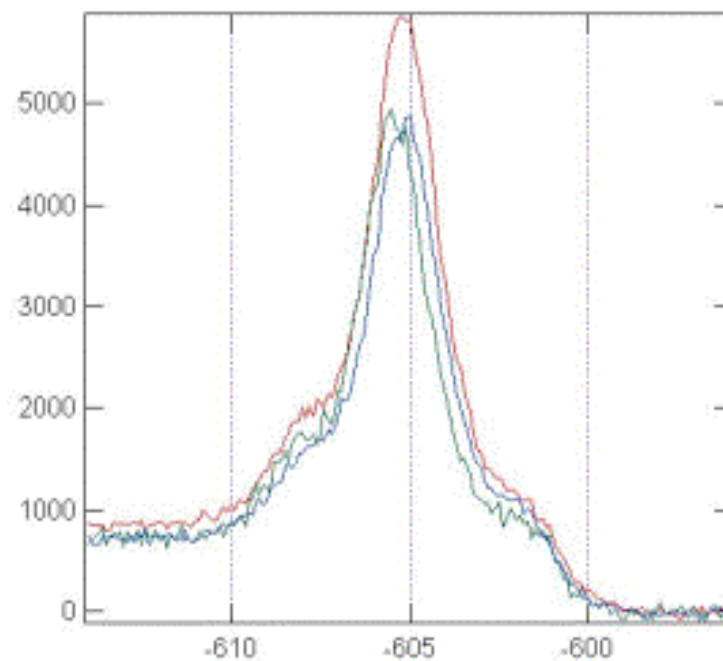
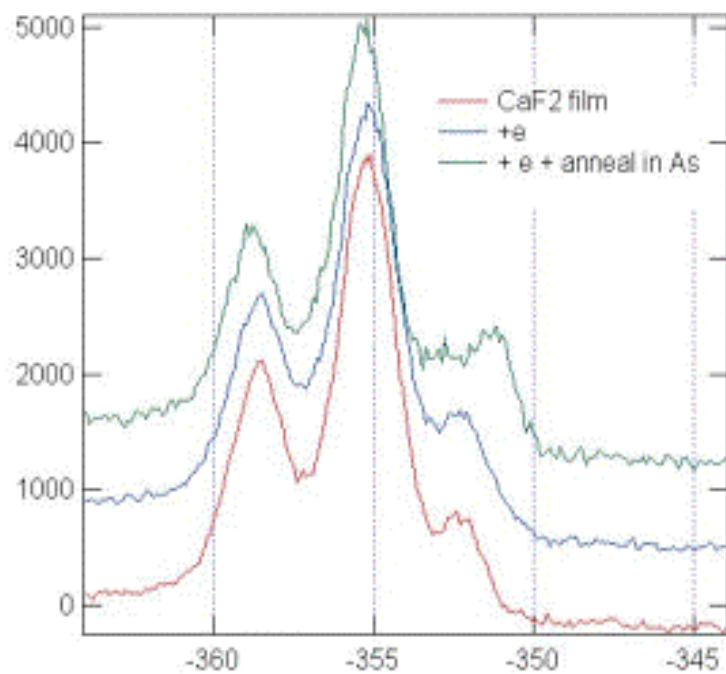


Knotek-Feibelman

- e^- of at least $\sim 40\text{eV}$ are “shot” at the sample
- A hole is created in the Ca 3p band
- An e^- from the F 2p decays into the hole
- An Auger e^- is ejected, leaving the F ion with a net $1+$ charge

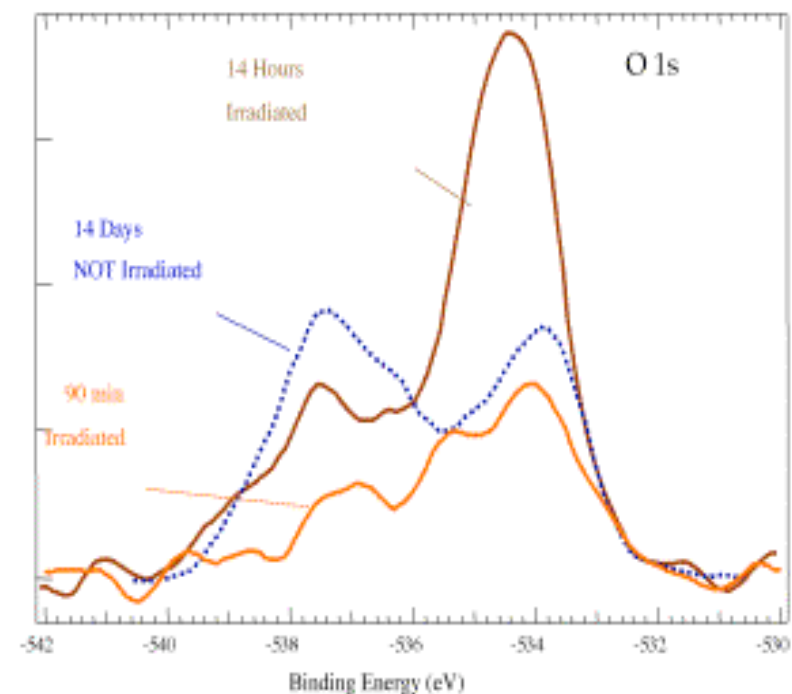
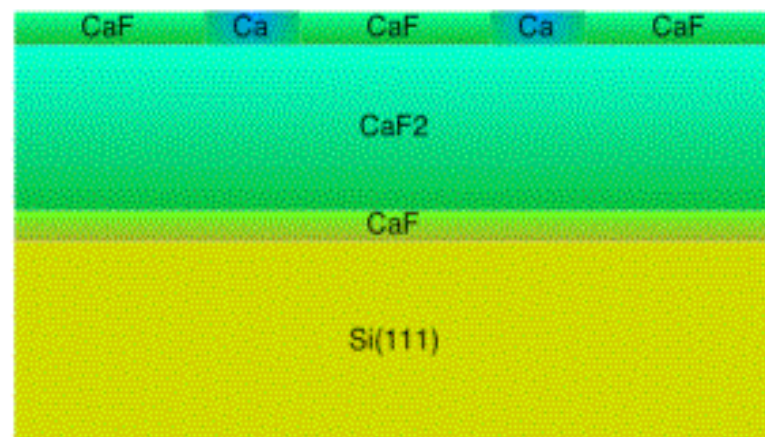


Post Irradiation



Surface of CaF_2 (post irradiation)

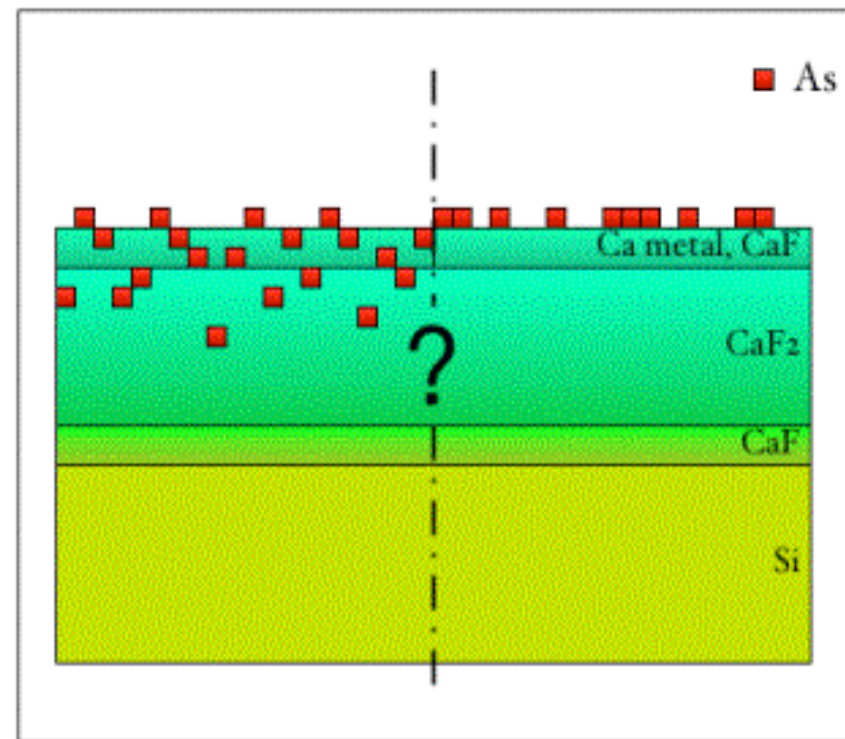
- Ca metal and CaF
- Very Reactive with residual gases in Chamber



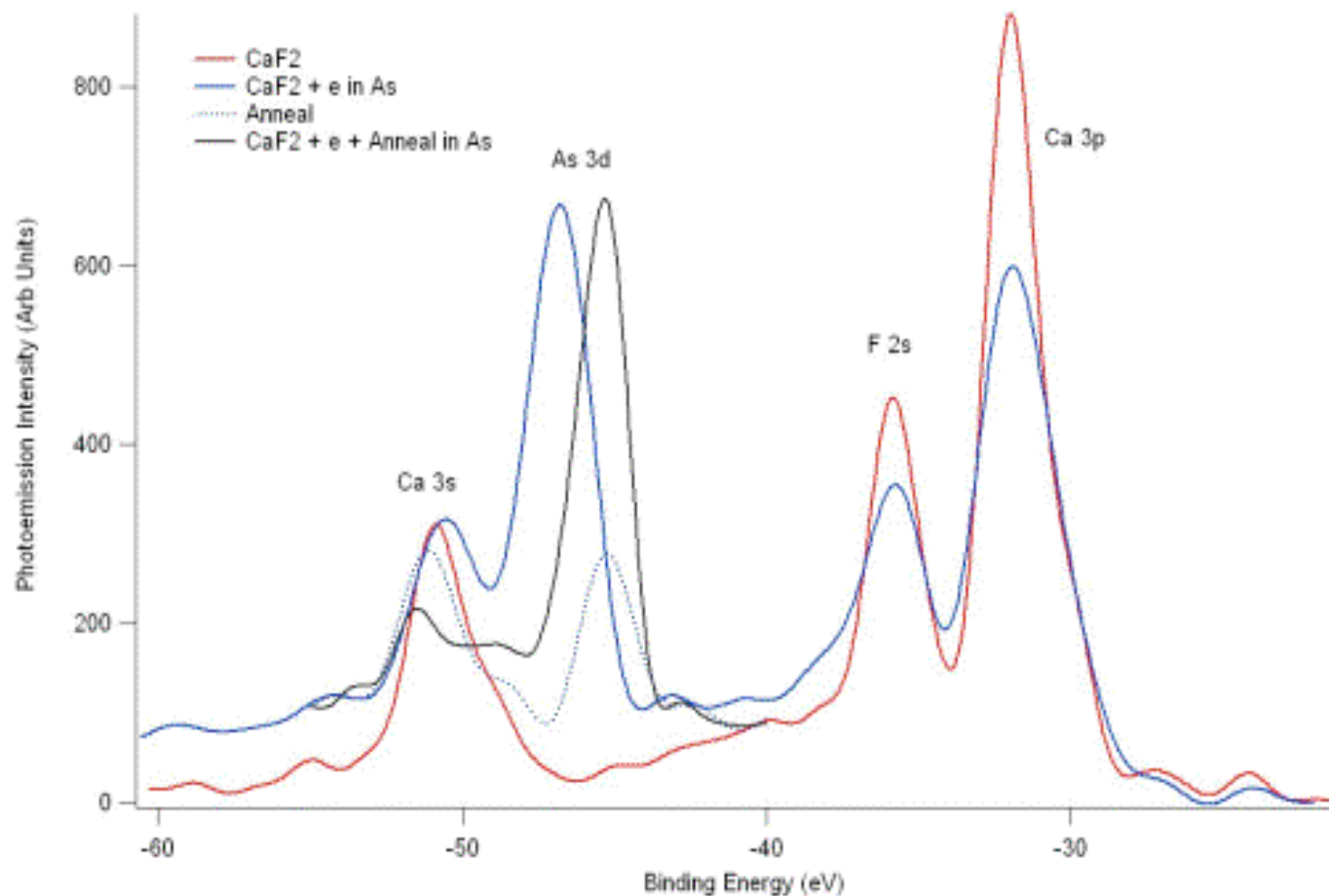
- Passivate Surface: As

The Question

- As works to passivate the surface
- Only sticks to Irradiated Surface
- Where is the As?
 - Surface, only $1/3$ ML?
 - Diffused into Crystal?

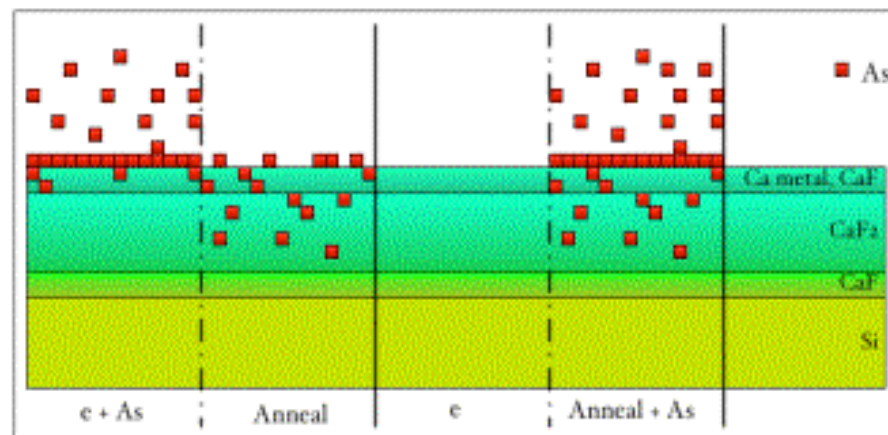


Irradiation with As



“The Answer”

- Irradiated CaF_2 allows for 1 ML of As to be deposited
- Annealing induces diffusion into the Crystal
- When Annealed in As, ML remains



Summary

- MBE allows for controllable growth of Thin CaF_2 Films
- XPS is a relatively fast way to analyze the composition of Thin Films
- ESD, while increasing the Surface energy, creates a highly reactive surface
- As for the As?
 - ML on surface diffuses down with annealing

Acknowledgements



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